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             7 S L8 AND L11
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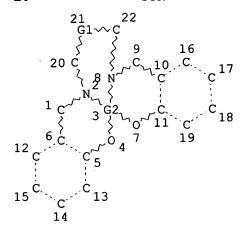
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STEREO ATTRIBUTES: NONE

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L3 (11715)SEA FILE=REGISTRY ABB=ON L2 AND (MN OR CO OR CU OR FE OR V
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CR OR NI)/ELS L4 STR



REP G1=(0-2) C
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DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 22

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L13
              7 S L8 AND L11
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                             COPYRIGHT 1999 ACS
L12 ANSWER 1 OF 29 CAPLUS
     1998:176006 CAPLUS
AN
                                                                 308-4491
DN
     128:226221
    Methods using manganese superoxide dismutase-deficient mouse for testing
ΤI
     compounds for use as therapeutic antioxidants
     Wallace, Douglas C.; Melov, Simon L.; Crapo, James D.; Day, Brian J.
IN
     Emory University, USA; Duke University
PΑ
SO
     PCT Int. Appl., 47 pp.
     CODEN: PIXXD2
DT
     Patent
LA
    English
FAN.CNT 1
     PATENT NO.
                      KIND
                            DATE
                                           APPLICATION NO.
                                                             DATE
     WO 98<u>10057</u>
ΡI
                      A1
                            19980312
                                           WO 97-US15814
                                                             19970905
         W: AU, CA, JP
         RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT,
SE
     AU 9742580
                       Α1
                            19980326
                                           AU 97-42580
                                                             19970905
PRAI US 96-24702
                      19960906
     WO 97-US15814
                      19970905
    Methods are described for the testing of compds. of potential usefulness
     as therapeutic antioxidants and/or as therapeutic free radical
     scavengers. The animal model for testing such compds. is the Sod2CJE
     homozygous manganese superoxide dismutase-deficient mouse. When pups of
     these mice are treated with certain antioxidants, they survive
     past about 7 days of age, and later develop characteristic histol.
changes
     and characteristic neurobehavioral disorders. Those treated mice can be
     further treated with test compds. Which may or may not cross the blood
     brain barrier, and the life span and phys. and neurobehavioral
```

```
characteristics of those mice provide information about the potential
     utility of the test compd. as a therapeutic antioxidant.
     Phenotypes of the treated mice allow conclusions regarding targeted areas
     of the brain and thus, applications to particular disorders, e.g.
     Parkinsonism.
     53177-12-1
IT
     RL: BAC (Biological activity or effector, except adverse); THU
     (Therapeutic use); BIOL (Biological study); USES (Uses)
        (manganese superoxide dismutase-deficient mouse for testing compds.
for
        use as therapeutic antioxidants)
     ICM C12N005-00
IC
     ICS C12N015-00; A61K049-00
     1-1 (Pharmacology)
CC
     Section cross-reference(s): 14, 78
     superoxide dismutase deficient mouse antioxidant effectiveness;
ST
     radical scavenger effectiveness MnSOD deficient mouse; Parkinsonism drug
     superoxide dismutase deficient mouse
TΤ
    Genes (animal)
     RL: BPR (Biological process); BIOL (Biological study); PROC (Process)
        (apoptotic; manganese superoxide dismutase-deficient mouse for testing
        compds. for use as therapeutic antioxidants)
ΙT
    Convulsions
        (audiogenic; manganese superoxide dismutase-deficient mouse for
testing
        compds. for use as therapeutic antioxidants)
IT
    Liver
        (lipids; manganese superoxide dismutase-deficient mouse for testing
        compds. for use as therapeutic antioxidants)
ΙT
     Hypolipemic agents
        (liver lipids; manganese superoxide dismutase-deficient mouse for
        testing compds. for use as therapeutic antioxidants)
IT
    Antioxidants
    Blood-brain barrier
    Brain injury
    Cardiomyopathy
    Cardioprotectants
     Drug screening
    Mitochondria
    Mouse
    Radical scavengers
        (manganese superoxide dismutase-deficient mouse for testing compds.
for
        use as therapeutic antioxidants)
IT
     Behavior (animal)
        (neurobehavior; manganese superoxide dismutase-deficient mouse for
        testing compds. for use as therapeutic antioxidants)
ΙT
     Genes (animal)
     RL: BPR (Biological process); BIOL (Biological study); PROC (Process)
        (sod2; manganese superoxide dismutase-deficient mouse for testing
        compds. for use as therapeutic antioxidants)
ΙT
                  73202-95-6
     53177-12-1
     RL: BAC (Biological activity or effector, except adverse); THU
     (Therapeutic use); BIOL (Biological study); USES (Uses)
        (manganese superoxide dismutase-deficient mouse for testing compds.
for
        use as therapeutic antioxidants)
```

IT 9054-89-1, Superoxide dismutase

RL: BPR (Biological process); BIOL (Biological study); PROC (Process) (manganese; manganese superoxide dismutase-deficient mouse for testing compds. for use as therapeutic antioxidants)

IT 11132-78-8, Manganese chloride 14609-54-2

RL: RCT (Reactant)

(reaction; manganese superoxide dismutase-deficient mouse for testing compds. for use as therapeutic **antioxidants**)

IT 53177-12-1

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (manganese superoxide dismutase-deficient mouse for testing compds.

for

use as therapeutic antioxidants)

RN 53177-12-1 CAPLUS

L12 ANSWER 2 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1997:189612 CAPLUS

DN 127:579

TI Vasodilatory effects of a salen-manganese complex with potent oxyradical scavenger activities

AU Barandier, Christine; Boucher, Francois; Malfroy, Bernard; De Leiris, Joel

CS Groupe Physiopathologie Cellulaire Cardiaque, Universite Joseph Fourier, Grenoble, F-38041, Fr.

SO J. Vasc. Res. (<u>1997</u>), 34(1), 49-57 CODEN: JVREE9; ISSN: 1018-1172

PB Karger

DT Journal

LA English

AB The effects of EUK-8, a salen-Mn complex with high superoxide dismutase and catalase activities, on rat aorta were investigated. EUK-8 protected the acetylcholine-induced relaxation of rat aortic rings from inhibition by superoxide anions and reduced H2O-induced relaxation. Moreover, EUK-8 dose-dependently relaxed rat aorta precontracted with phenylephrine and decreased the vascular tone of noncontracted aortic rings. The relaxant effect of EUK-8 was potentiated by endothelium abrasion and/or

preincubation with N-nitro-L-arginine Me ester, an inhibitor of NO synthase. Indomethacin had no effect on the action of EUK-8, showing

that

it was not dependent on prostacyclin synthesis. Methylene blue, an inhibitor of sol. guanylate cyclase, partly abolished relaxation induced by EUK-8. Incubation of rat aorta with EUK-8 induced an increase in vascular cAMP content. The lack of inhibition by dl-propranolol showed that adenylate cyclase activation by EUK-8 was not mediated through .beta.-adrenergic receptors. The inhibition of the effects of EUK-8 by tetraethylammonium and glibenclamide showed the implication of potassium channels in the intracellular cascade triggered by EUK-8. The vasorelaxant activity of EUK-8 was neither affected by xanthine oxidase inhibition nor by superoxide anion scavenging. Finally, the ligand for EUK-8 without its antioxidant activities because of the absence of manganese, conversely potentiated phenylephrine-induced contraction of aortic rings.

IT 53177-12-1, EUK-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (vasodilatory effects of a salen-manganese complex with potent oxyradical scavenger activities)

CC 1-8 (Pharmacology)

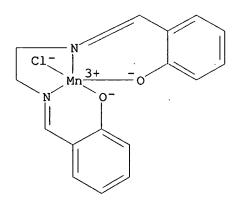
IT 53177-12-1, EUK-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (vasodilatory effects of a salen-manganese complex with potent oxyradical scavenger activities)

IT **53177-12-1**, EUK-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (vasodilatory effects of a salen-manganese complex with potent oxyradical scavenger activities)

RN 53177-12-1 CAPLUS



L12 ANSWER 3 OF 29 CAPLUS COPYRIGHT 1999 ACS

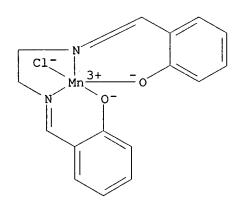
AN 1997:30121 CAPLUS

DN 126:126681

```
ΤI
     EUK-134, a synthetic superoxide dismutase and catalase mimetic, protects
     rat kidneys from ischemia-reperfusion-induced damage
     Gianello, Pierre; Saliez, Alain; Bufkens, Xavier; Pettinger, Rene;
ΑU
     Misseleyn, Dominique; Hori, Seiji; Malfroy, Bernard
     Medical School, University Louvain, Brussels, 1200, Belg.
CS
     Transplantation (1996), 62(11), 1664-1666
SO
     CODEN: TRPLAU; ISSN: 0041-1337
PB
     Williams & Wilkins
DT
     Journal
     English
LA
AB
     The effect of a new synthetic superoxide dismutase and catalase mimetic
     was investigated on renal ischemia-reperfusion syndrome in rats.
     Synthetic salen-manganese complexes have characteristics that might
     facilitate their potential usefulness as therapeutic agents: (1) unlike
     proteinaceous antioxidant enzymes, synthetic complexes, due to
     their low mol. wt., have a better stability and bioavailability; (2) they
     have a catalytic activity enhancing their efficiency over noncatalytic
     reactive oxygen metabolite scavengers; and finally, (3) exhibiting
     combined superoxide dismutase and catalase activity, they destroy both
     superoxide anions and hydrogen peroxides, thereby enhancing their
     protective effect on ischem. injured tissues. One such compd., EUK-134,
     was tested in uninephrectomized rats that underwent a left renal artery
     clamping. After a 75-min left renal artery clamping, a single i.v.
     injection of EUK-134 at 0.2 mg/kg, just before unclamping, provided
     significantly better renal function recovery during the week after the
     ischemic insult compared with recovery of untreated animals.
     after several periods of renal ischemia (30, 45, 60, and 75 min of left
     renal artery clamping), EUK-134 given at a similar dose significantly
     improved the glomerular filtration rate after an acute ischemia of 30 and
     45 min, as assessed by EDTA 51Cr. Overall, these results show that
     synthetic superoxide dismutase-catalase mimetics such as EUK-134 can
     protect ischem. injured rat kidneys from ischemia-reperfusion syndrome
     when administered just before reperfusion.
TΤ
     186467-37-8, EUK 134
     RL: BAC (Biological activity or effector, except adverse); THU
      (Therapeutic use); BIOL (Biological study); USES (Uses)
         (EUK-134, synthetic superoxide dismutase and catalase mimetic,
protects
        rat kidneys from ischemia-reperfusion-induced damage)
CC
     1-8 (Pharmacology)
IT
     186467-37-8, EUK 134
     RL: BAC (Biological activity or effector, except adverse); THU
      (Therapeutic use); BIOL (Biological study); USES (Uses)
         (EUK-134, synthetic superoxide dismutase and catalase mimetic,
protects
        rat kidneys from ischemia-reperfusion-induced damage)
     186467-37-8, EUK 134
     RL: BAC (Biological activity or effector, except adverse); THU
      (Therapeutic use); BIOL (Biological study); USES (Uses)
         (EUK-134, synthetic superoxide dismutase and catalase mimetic,
protects
        rat kidneys from ischemia-reperfusion-induced damage)
     186467-37-8 CAPLUS
L12
     ANSWER 4 OF 29 CAPLUS COPYRIGHT 1999 ACS
AN
     1996:684323 CAPLUS
 DN
     126:17317
```

```
Free radicals in reperfusion-induced arrhythmias: study with EUK 8, a
TI
     novel nonprotein catalytic antioxidant
     Tanguy, Stephane; Boucher, Francois R.; Malfroy, Bernard; De Leiris, Joel
ΑU
     Physiopathologie Cellulaire Cardiaque, Universite Joseph Fourier,
CS
     Grenoble, Fr.
     Free Radical Biol. Med. (1996), 21(7), 945-954
SO
     CODEN: FRBMEH; ISSN: 0891-5849
PB
     Elsevier
DT
     Journal
     English
LA
     Oxyradicals have been implicated as a possible cause of postischemic
AB
     reperfusion arrhythmias (RA). However, the ability of enzymic scavengers
     such as superoxide dismutase and/or catalase to reduce RA remains
     controversial. The purpose of the present work was to det. whether a
     nonprotein catalytic antioxidant, EUK 8, may limit RA in
     isolated heart prepns. The catalytic dismutation of H2O2 by EUK 8 was
     demonstrated using a Clark electrode. EUK 8's ability to scavenge
     oxyradicals was studied in vitro by ESR in presence of superoxide-anion
     generating system. ESR concn.-effect curves obtained led the authors to
     use EUK 8 at 50 .mu.mol/l in isolated heart prepns. Isolated rat hearts
     were submitted to 10 min regional ischemia induced by left coronary
     ligation. Reperfusion was achieved by releasing the coronary ligation,
     and the incidence and duration of early ventricular arrhythmias were then
     investigated. In the treated-group, EUK 8 was added to the perfusion
     fluid (50 .mu.mol) 90 s before reperfusion. The results show that EUK 8
     reduced the severity of RA as assessed by the arrhythmia score
     measurement. In conclusion, EUK 8 is able to limit RA in the authors'
     expt. model. This effect might be related to the catalytic
     antioxidant properties of this complex.
TΤ
     53177-12-1, EUK 8
     RL: BUU (Biological use, unclassified); THU (Therapeutic use); BIOL
     (Biological study); USES (Uses)
        (EUK 8, a nonprotein catalytic antioxidant, limits
        reperfusion arrhythmias, which may be related to EUK 8's
      antioxidant properties)
CC
     14-5 (Mammalian Pathological Biochemistry)
     Section cross-reference(s): 1
IT
    Antioxidants
     Myocardial ischemia
     Oxidative stress (biological)
        (EUK 8, a nonprotein catalytic antioxidant, limits
        reperfusion arrhythmias, which may be related to EUK 8's
      antioxidant properties)
ΙT
     Reactive oxygen species
     RL: ADV (Adverse effect, including toxicity); BIOL (Biological study)
        (EUK 8, a nonprotein catalytic antioxidant, limits
        reperfusion arrhythmias, which may be related to EUK 8's
      antioxidant properties)
ΙT
     Reperfusion
        (of ischemic heart; EUK 8, a nonprotein catalytic antioxidant
        , limits reperfusion arrhythmias, which may be related to EUK 8's
      antioxidant properties)
     Ventricular arrhythmia
        (reperfusion-induced; EUK 8, a nonprotein catalytic antioxidant
        , limits reperfusion arrhythmias, which may be related to EUK 8's
```

antioxidant properties) IT 7722-84-1, Hydrogen peroxide, biological studies 7782-44-7D, Oxygen, 11062-77-4, Superoxide radicals RL: ADV (Adverse effect, including toxicity); BIOL (Biological study) (EUK 8, a nonprotein catalytic antioxidant, limits reperfusion arrhythmias, which may be related to EUK 8's antioxidant properties) 9001-05-2, Catalase ΙT RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (EUK 8, a nonprotein catalytic antioxidant, limits reperfusion arrhythmias, which may be related to EUK 8's antioxidant properties) 53177-12-1, EUK 8 ΙT RL: BUU (Biological use, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (EUK 8, a nonprotein catalytic antioxidant, limits reperfusion arrhythmias, which may be related to EUK 8's antioxidant properties) IΤ 53177-12-1, EUK 8 RL: BUU (Biological use, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (EUK 8, a nonprotein catalytic antioxidant, limits reperfusion arrhythmias, which may be related to EUK 8's antioxidant properties) 53177-12-1 CAPLUS RN Manganese, chloro[[2,2'-[1,2-ethanediylbis[(nitrilo-CN .kappa.N)methylidyne]]bis[phenolato-.kappa.O]](2-)]-, (SP-5-13)- (9CI) (CA INDEX NAME)



L12

AN 1996:316370 CAPLUS
DN 125:51912
TI Antioxidative activity of biologically active compounds:
 Measurement by Cypridina chemiluminescence method
AU Suzuki, N.; Mashiko, S.; Hamada, M.; Nomoto, T.; Hasegaga, M.; Yoda, B.
CS National University Fisheries, Shimonoseki, 759-65, Japan
SO Biolumin. Chemilumin., Proc. Int. Symp., 8th (1994), 219-222. Editor(s):
 Campbell, Andrew Keith; Kricka, Larry J.; Stanley, Philip E. Publisher:
 Wiley, Chichester, UK.

ANSWER 5 OF 29 CAPLUS COPYRIGHT 1999 ACS

```
CODEN: 62UZAR
DT
     Conference
LA
     English
     The highly sensitive Cypridina chemiluminescence method previously
AB
     developed by the authors was used to det. the antioxidative
     activity of various peptides and salcomine derivs. Proteins from marine
     life showed 1-2 orders larger reaction rate consts. than did those from
     land animals and plants. Hydrolyzates of the proteins from land animals
     showed larger consts. than did the unhydrolyzed proteins. The salcomine
     derivs. were also strong antioxidants.
     14167-18-1, Salcomine 14167-18-1D, Salcomine, derivs.
IT
     14167-20-5 60306-16-3
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (antioxidative activity of biol. active compds.)
     6-7 (General Biochemistry)
CC
     antioxidant protein salcomine deriv
ST
IT
     Polydactylus sexfilis
        (actin and actomyosin; antioxidative activity of biol. active
        compds.)
IT
     Antioxidants
        (antioxidative activity of biol. active compds.)
TΤ
     Protein hydrolyzates
     Caseins, biological studies
     Peptides, biological studies
     Proteins, biological studies
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (antioxidative activity of biol. active compds.)
IT
     Euthynnus pelamis
        (collagen and gelatin; antioxidative activity of biol. active
        compds.)
IT
     Proteins, specific or class
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (egg white K; antioxidative activity of biol. active compds.)
IT
     Sardine
        (myofibril; antioxidative activity of biol. active compds.)
IT
     Soybean
        (peptides; antioxidative activity of biol. active compds.)
ΙT
     Myosins
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (red horsehead; antioxidative activity of biol. active
        compds.)
     Collagens, biological studies
IT
     Gelatins, biological studies
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (skipjack tuna; antioxidative activity of biol. active
        compds.)
ΙT
     Actins
     Actomyosins
     RL: BAC (Biological activity or effector, except adverse); BIOL
     (Biological study)
        (threadfin bream; antioxidative activity of biol. active
        compds.)
```

IT Gelatins, biological studies RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (hydrolyzates, skipjack tuna; antioxidative activity of biol. active compds.) ΙT Organelle (myofibril, sardine; antioxidative activity of biol. active compds.) ΙT 14167-18-1, Salcomine 14167-18-1D, Salcomine, derivs. 41139-17-7 60306-16-3 14167-20-5 RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (antioxidative activity of biol. active compds.) 14167-18-1, Salcomine 14167-18-1D, Salcomine, derivs. ΙT 14167-20-5 60306-16-3 RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (antioxidative activity of biol. active compds.) RN 14167-18-1 CAPLUS Cobalt, CN [[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

14167-18-1 CAPLUS

RN

CN Cobalt,
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 14167-20-5 CAPLUS

CN Nickel,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 60306-16-3 CAPLUS

CN Cobalt, [[2,2'-[(1,1,2,2-tetramethyl-1,2-ethanediyl)bis[(nitrilo-.kappa.N)methylidyne]]bis[phenolato-.kappa.O]](2-)]-, (SP-4-2)- (9CI)

(CA INDEX NAME)

L12 ANSWER 6 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1996:180742 CAPLUS

DN 124:279072

TI .beta.-Amyloid toxicity in organotypic hippocampal cultures: protection

by

EUK-8, a synthetic catalytic free radical scavenger

AU Bruce, Annadora J.; Malfroy, Bernard; Baudry, Michel

CS Neurosci. Program, Univ. Southern California, Los Angeles, CA, 90089-2520,

USA

SO Proc. Natl. Acad. Sci. U. S. A. (1996), 93(6), 2312-16 CODEN: PNASA6; ISSN: 0027-8424

DT Journal

LA English

AB Oxygen free radicals have been proposed to mediate amyloid peptide (.beta.AP)-induced neurotoxicity. To test this hypothesis, we evaluated the effect of EUK-8, a synthetic catalytic superoxide and hydrogen peroxide scavenger, on neuronal injury produced by .beta.AP in organotypic

hippocampal slice cultures. Cultures of equiv. postnatal day 35 (defined as mature) and 14 (defined as immature) were exposed to various concns.

of

.beta.AP (1-42 or 1-40) in the absence or presence of 25 .mu.M EUK-8 for up to 72 h. Neuronal injury was assessed by lactate dehydrogenase release

and semiquant. anal. of propidium iodide uptake at various times after the $% \left(1\right) =\left(1\right) +\left(1\right)$

initiation of .beta.AP exposure. Free radical prodn. was inferred from the relative increase in dichlorofluorescein fluorescence, and the degree of lipid peroxidn. was detd. by assaying thiobarbituric acid-reactive substances. Treatment of mature cultures with .beta.AP (50-250 .mu.g/mL) in serum-free conditions resulted in a reproducible pattern of damage, causing a time-dependent increase in neuronal injury accompanied with formation of reactive oxygen species. However, immature cultures were entirely resistant to .beta.AP-induced neurotoxicity and also

demonstrated

no dichlorofluorescein fluorescence or increased lipid peroxidn. after .beta.AP treatment. Moreover, mature slices exposed to .beta.AP in the presence of 25 .mu.M EUK-8 were significantly protected from .beta.AP-induced neurotoxicity. EUK-8 also completely blocked

.beta.AP-induced free radical accumulation and lipid peroxidn. These results not only support a role for oxygen free radicals in .beta.AP toxicity but also highlight the therapeutic potential of synthetic radical

scavengers in Alzheimer disease.

IT 53177-12-1, Euk-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(.beta.-amyloid toxicity in organotypic hippocampal cultures inhibition

by EUK-8, a synthetic catalytic free radical scavenger)

CC 1-12 (Pharmacology)

Section cross-reference(s): 14

ST antioxidant EUK8 beta amyloid neurotoxicity hippocampus

IT Antioxidants

Nerve

(.beta.-amyloid toxicity in organotypic hippocampal cultures inhibition $% \left(\frac{1}{2}\right) =\frac{1}{2}\left(\frac{1}{2}\right) +\frac{1}{2}\left(\frac{1}{2}\right) +\frac{1$

by EUK-8, a synthetic catalytic free radical scavenger)

IT **53177-12-1**, Euk-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(.beta.-amyloid toxicity in organotypic hippocampal cultures inhibition

by EUK-8, a synthetic catalytic free radical scavenger)

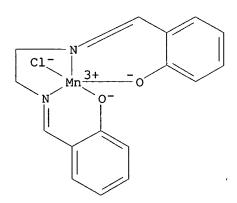
IT **53177-12-1**, Euk-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

 $\hbox{(.beta.-amyloid toxicity in organotypic hippocampal cultures inhibition}\\$

by EUK-8, a synthetic catalytic free radical scavenger)

RN 53177-12-1 CAPLUS



L12 ANSWER 7 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1995:994554 CAPLUS

DN 124:55795

TI Preparation and formulation of chromanylideneaminoguanidines and analogs
Page 15

as Maillard reaction-inhibiting antioxidants

IN Ohuchida, Shuichi; Hasegawa, Tomoyuki; Kishimoto, Kazuo

PA Ono Pharmaceutical Co., Ltd., Japan

SO PCT Int. Appl., 90 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

----WO 9523796 A1 19950908 WO 95-JP294 19950227

PI WO 9523796 A1 19950908 WO 95-JP W: CA, CN, JP, KR, US

RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE

PRAI JP 94-55223 19940301

OS MARPAT 124:55795

GI

$$Q1= \begin{array}{c} (CH_2)_n \\ R7 \\ O \end{array} \qquad (R5)_m$$

AB The title compds. R1CR2(:N)NHC(:NH)NHR3 (I) [R1 represents R4A (A is a single bond, alkylene or phenylalkylene; and R4 is Q1); R2 represents hydrogen, alkyl, Ph, phenylalkyl, etc.; or R1 and R2 together with the carbon atom to which they are bonded represent cycloalkyl fused with R4; and R3 represents hydrogen, alkyl or acyl; provided when A represents phenylalkylene, the Ph is bonded to the carbon atom to which R2 is bonded;

ΙI

R5 = H, alkyl; R6 = H, alkyl, acyl; R7 = H, alkyl, etc.; m = 1 - 9; n = 0 - 2] are prepd. I are useful in the treatment and prevention of complications of diabetes and aging. The title compd. II was prepd. in a multistep process starting with 2,5-dimethylbenzoquinone. II in vitro showed IC50 of 7.3 .mu.M against lipid peroxidn.

IT 14167-18-1, Salcomine

RL: CAT (Catalyst use); USES (Uses)

(prepn. of chromanylideneaminoguanidines and analogs as Maillard reaction-inhibiting antioxidants)

IC ICM C07D307-79

ICS C07D311-72; C07D405-06

ICA A61K031-34; A61K031-35; A61K031-44

CC 27-14 (Heterocyclic Compounds (One Hetero Atom))
 Section cross-reference(s): 1, 63

ST chromanylideneaminoguanidine prepn Maillard reaction inhibitor

```
antioxidant; diabetes complication treatment
     chromanylideneaminoguanidine prepn; aging treatment
     chromanylideneaminoguanidine prepn
    Antioxidants
ΤT
    Maillard reaction
        (prepn. of chromanylideneaminoguanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
IT
     171967-93-4P
                    171967-94-5P
                                   171967-95-6P
                                                  171967-96-7P
                                                                 171967-97-8P
     171967-98-9P
                    171967-99-0P
                                   171968-00-6P
                                                  171968-01-7P
                                                                 171968-02-8P
    171968-03-9P
                    171968-04-0P
                                   171968-05-1P
                                                  171968-06-2P
                                                                 171968-07-3P
    171968-08-4P
                    171968-09-5P
                                   171968-10-8P
                                                  171968-11-9P
                                                                 171968-12-0P
    RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic
    preparation); THU (Therapeutic use); BIOL (Biological study); PREP
     (Preparation); USES (Uses)
        (prepn. of chromanylideneaminoquanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
ΙT
    14167-18-1, Salcomine
    RL: CAT (Catalyst use); USES (Uses)
        (prepn. of chromanylideneaminoguanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
                                         95-87-4, 2,5-Dimethylphenol
     74-88-4, Methyl iodide, reactions
IT
     107-30-2, Chloromethyl methyl ether
                                          108-24-7, Acetic anhydride
    108-30-5, Succinic anhydride, reactions
                                               124-40-3, Dimethylamine,
                 137-18-8, 2,5-Dimethylbenzoquinone
    reactions
                                                      150-78-7,
                           526-86-3, 2,3-Dimethylbenzoquinone
    1,4-Dimethoxybenzene
                                                                 527-61-7,
                                556-82-1, 3-Methyl-2-buten-1-ol
     3,5-Dimethylbenzoquinone
                                                                 563-47-3,
     3-Chloro-2-methylpropene
                                867-13-0, Ethyl diethylphosphonoacetate
    1253-46-9, (4-Methoxycarbonylbenzyl)triphenylphosphonium bromide
    1937-19-5, Aminoquanidine hydrochloride
                                               2605-67-6, Methyl
     (triphenylphosphoranylidene)acetate
                                           3282-30-2, Pivaloyl chloride
     4885-02-3, .alpha.,.alpha.-Dichloromethyl methyl ether
                                                              171968-46-0
    RL: RCT (Reactant)
        (prepn. of chromanylideneaminoguanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
                              1077-69-6P
                                           1083-11-0P
    615-90-7P
                1015-55-0P
                                                        1084-74-8P
6133-18-2P
                                               74785-10-7P
                                                            78707-88-7P
    19206-87-2P
                   26172-17-8P
                                 60404-99-1P
    149467-89-0P
                                                  162963-41-9P
                    162963-37-3P
                                   162963-39-5P
                                                                 170728-04-8P
                    171968-14-2P
                                   171968-15-3P
                                                  171968-16-4P
                                                                 171968-17-5P
    171968-13-1P
    171968-18-6P
                    171968-19-7P
                                   171968-20-0P
                                                  171968-21-1P
                                                                 171968-22-2P
                                                                 171968-27-7P
    171968-23-3P
                    171968-24-4P
                                   171968-25-5P
                                                  171968-26-6P
                    171968-29-9P
    171968-28-8P
                                   171968-30-2P
                                                  171968-31-3P
                                                                 171968-32-4P
    171968-33-5P
                    171968-34-6P
                                   171968-35-7P
                                                  171968-36-8P
                                                                 171968-37-9P
    171968-38-0P
                    171968-39-1P
                                   171968-40-4P
                                                  171968-41-5P
                                                                  171968-42-6P
    171968-43-7P
                    171968-44-8P
                                   171968-45-9P
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of chromanylideneaminoguanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
IT
    14167-18-1, Salcomine
    RL: CAT (Catalyst use); USES (Uses)
        (prepn. of chromanylideneaminoguanidines and analogs as Maillard
        reaction-inhibiting antioxidants)
RN
     14167-18-1 CAPLUS
    Cobalt,
CN
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.0]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
```

hydrophobic

```
L12 ANSWER 8 OF 29 CAPLUS COPYRIGHT 1999 ACS
     1995:708425 CAPLUS
ΑN
DN
     123:85859
     Method for improving the stability of dyeings on hydrophobic fibers, its
TΙ
     use on polyamide and polyester fibers and treated textiles from
     Schlenker, Wolfgang; Strahm, Ulrich; Fuso, Francesco
IN
     Ciba-Geigy A.-G., Switz.
PA
SO
     Ger. Offen., 12 pp.
     CODEN: GWXXBX
DT
     Patent
LA
     German
FAN.CNT 1
                      KIND
     PATENT NO.
                            DATE
                                           APPLICATION NO.
                                                             DATE
                            19950302
                                           DE 94-4429470
                                                             19940819
    DE 4429470
                       A1
PΙ
                      19930823
PRAI CH 93-2500
os
    MARPAT 123:85859
AB
     Title method comprises treating the dyed textile with a fiber stabilizer,
     e.g., antioxidants or light stabilizers, in supercrit. CO2. A
     striped polyester textile was treated with 2,4-diphenyl-6-(2-hydroxy-4-
     methoxyphenyl)-1,3,5-triazine (I) in supercrit. CO2 in an autoclave at
     130.degree. and 250 bar for 30 min to give 75% exhaustion of I.
IT
     14167-15-8
     RL: TEM (Technical or engineered material use); USES (Uses)
        (light stabilizer; in improving the stability of dyeings on
hydrophobic
        fibers)
IC
     ICM D06P005-04
ICA
     D06P005-06; D06P005-10
CC
     40-6 (Textiles and Fibers)
ST
     light stabilizer supercrit carbon dioxide textile; dyeing stability
     supercrit carbon dioxide; antioxidant supercrit carbon dioxide
     textile dyeing; polyester fiber dyeing stability; polyamide fiber dyeing
     stability
IT
     Antioxidants
     Light stabilizers
        (in improving the stability of dyeings on hydrophobic fibers)
     14167-15-8
                  52829-07-9
                               106556-36-9
     RL: TEM (Technical or engineered material use); USES (Uses)
        (light stabilizer; in improving the stability of dyeings on
```

Page 18

fibers)

IT 14167-15-8

RL: TEM (Technical or engineered material use); USES (Uses) (light stabilizer; in improving the stability of dyeings on hydrophobic

fibers)

RN 14167-15-8 CAPLUS

CN Copper,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

L12 ANSWER 9 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1995:209409 CAPLUS

DN 122:46423

TI Preparation and characterization of Mn-salophen complex with superoxide scavenging activity

AU Liu, Zheng-Xian; Robinson, Gina B.; Gregory, Eugene M.

CS Department of Biochemistry and Anaerobic Microbiology, Virginia Polytechnic Institute, VA, 24061-0308, USA

SO Arch. Biochem. Biophys. (1994), 315(1), 74-81 CODEN: ABBIA4; ISSN: 0003-9861

DT Journal

LA English

AB Mn(III)-salophen complex with superoxide scavenging activity was prepd. from manganese(III) acetate dihydrate and salophen in ethanol. Visible absorption spectrum of the red-brown complex exhibits a shoulder at 430 nm

which was absent with either salophen or manganic acetate alone. Titrn. of salophen with manganese(III) is consistent with a 1:1 Mn to salophen stoichiometry of the complex based on changes in the absorbance at 500 nm or of superoxide scavenging activity. The superoxide dismutase

(SOD)-like

activity of the complex in the xanthine-xanthine oxidase/cytochrome c assay was 1450 units/mg salophen. The SOD activity of the complex was suppressed 50% in the presence of EDTA (1 mM), but was not altered in the presence of bovine serum albumin (1 mg/mL) or crude protein ext. of Escherichia coli QC779 sodA-sodB- (1 mg/mL). E. coli QC779 sodA-sodB- grew scantily after an 8-h lag phase in aerobic M63 glucose minimal medium. The aerobic growth of the E. coli SOD double mutant in glucose minimal medium was greatly enhanced in the presence of 5 or 10 .mu.M

Mn-salophen complex compared to that of control after 24 h incubation. Mn-desferal green complex (10 .mu.M) and pink complex (5 .mu.M) also increased growth rate of E. coli QC779 sodA-sodB- but to a lesser extent than Mn-salophen complex. However, the growth was completely inhibited

by

50 .mu.M Mn-salophen complex, 100 .mu.M Mn-desferal green complex, or 10 .mu.M Mn-desferal pink complex.

IT 100183-26-4P

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(prepn. and characterization of Mn-salophen complex with superoxide scavenging activity)

CC 1-12 (Pharmacology)

IT Antioxidants

(prepn. and characterization of Mn-salophen complex with superoxide scavenging activity)

IT 100183-26-4P

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(prepn. and characterization of Mn-salophen complex with superoxide scavenging activity)

IT 100183-26-4P

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(prepn. and characterization of Mn-salophen complex with superoxide scavenging activity)

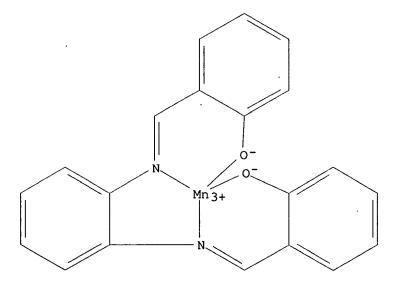
RN 100183-26-4 CAPLUS

CN Manganese(1+), [[2,2'-[1,2-phenylenebis[(nitrilo-

.kappa.N)methylidyne]]bis[phenolato-.kappa.O]](2-)]-, (SP-4-2)- (9CI)

(CA

INDEX NAME)



L12 ANSWER 10 OF 29 CAPLUS COPYRIGHT 1999 ACS AN 1995:41949 CAPLUS

```
122:23297
DN
     Antioxidant activity of chlorophyll derivatives: measurement by
ΤI
     Cypridina chemiluminescence method.
     Suzuki, N.; Nakamura, K.; Namiki, M.; Nomoto, T.; Yoda, B.; Saeki, A.
ΑU
     Shimonoseki Univ. Fisheries, Yoshimi, 759-65, Japan
CS
     Chem. Funct. Dyes, Proc. Int. Symp., 2nd (1993), Meeting Date 1992,
SO
130-5.
     Editor(s): Yoshida, Z.; Shirota, Y. Publisher: Mita Press, Tokyo, Japan.
     CODEN: 59TQAX
DT
     Conference
LA
     English
AΒ
     Several metallochlorophyllins and chlorins showed antioxidant
     activity to superoxide in aq. soln., as detd. by the Cypridina
     chemiluminescence method (Suzuki, N., et al., 1991). The compds. are
     potential agents for oxidative stress relief in cancer patients.
ΙT
     14167-18-1, Salcomin
     RL: BAC (Biological activity or effector, except adverse); THU
     (Therapeutic use); BIOL (Biological study); USES (Uses)
        (antioxidant activity of chlorophyll derivs. to superoxide)
CC
     1-6 (Pharmacology)
     antioxidant chlorophyll deriv oxidative stress cancer
ST
ΙT
     Antioxidants
        (antioxidant activity of chlorophyll derivs. to superoxide)
ΙT
     Neoplasm inhibitors
        (chlorophyll deriv. antioxidants for oxidative stress relief)
     11006-34-1 14167-18-1, Salcomin 69138-22-3
TΤ
                                                    72984-36-2
                                 159602-67-2
                                               159602-68-3
     100111-78-2
                   152695-46-0
                                                            159602-69-4
     159602-70-7
     RL: BAC (Biological activity or effector, except adverse); THU
     (Therapeutic use); BIOL (Biological study); USES (Uses)
        (antioxidant activity of chlorophyll derivs. to superoxide)
IT
     14167-18-1, Salcomin
     RL: BAC (Biological activity or effector, except adverse); THU
     (Therapeutic use); BIOL (Biological study); USES (Uses)
        (antioxidant activity of chlorophyll derivs. to superoxide)
     14167-18-1 CAPLUS
RN
     Cobalt,
CN
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
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L12 ANSWER 11 OF 29 CAPLUS COPYRIGHT 1999 ACS
     1994:511398 CAPLUS
AN
DN
     121:111398
     Polyolefin stabilization against UV light
TI
     Walters, John Phillip
IN
PA
     Phillips Petroleum Co., USA
     Eur. Pat. Appl., 30 pp.
SO
     CODEN: EPXXDW
DT
     Patent
     English
LA
FAN.CNT 1
     PATENT NO.
                      KIND
                            DATE
                                           APPLICATION NO.
                      ____
                            -----
                                           ------
     EP 567117
                      A1
                            19931027
                                           EP 93-106550
                                                            19930422
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, LU, NL, PT, SE
     US 5310771
                            19940510
                                           US 92-872863
                                                            19920423
                       Α
     JP 06016875
                       A2
                            19940125
                                           JP 93-96190
                                                            19930422
PRAI US 92-872863
                      19920423
     MARPAT 121:111398
OS
     Polyolefins, esp. pigmented fibers, are stabilized against UV light by
     incorporation of a hindered amine, a metal phosphonate, and optionally, a
     phenolic antioxidant and an org. phosphite. Thus, polypropylene
     fibers contg. C.I. Pigment Blue 15:1, Irgastab 2002 (I, a Ni
phosphonate),
     and GR-3150 [2,4,6-tris[N-cyclohexyl-N-[2-(3,3,5,5-
     tetramethylpiperazinonyl)ethyl]]-s-triazine] exhibited significantly less
     degrdn. in fadometer at 89.degree. and relative humidity 50% than similar
     fibers not contg. I.
ΙT
     20437-10-9, C.I. Pigment Orange 65
     RL: USES (Uses)
        (polypropylene fibers pigmented by, light stabilizers for, hindered
        amine-metal phosphonate mixts. as)
IC
     ICM C08L023-10
     ICS C08K005-00
ICI C08K005-00, C08K005-3462, C08K005-3435, C08K005-3492, C08K005-5333,
     C08K005-5393
CC
     40-3 (Textiles and Fibers)
     Section cross-reference(s): 37
     81-33-4, C.I. Pigment Violet 29
IT
                                       147-14-8, C.I. Pigment Blue 15:1
     980-26-7, C.I. Pigment Red 122
                                      1047-16-1, C.I. Pigment Violet 19
     1309-37-1, C.I. Pigment Red 101, uses
                                           1328-53-6, C.I. Pigment Green 7
     1344-37-2, C.I. Pigment Yellow 34
                                         5280-78-4, C.I. Pigment Red 144
     5521-31-3, C.I. Pigment Red 179
                                       5590-18-1, C.I. Pigment Yellow 110
     6358-30-1, C.I. Pigment Violet 23 12656-85-8, C.I. Pigment Red 104
     13463-67-7, C.I. Pigment White 6, uses 20437-10-9, C.I. Pigment
     Orange 65
     RL: USES (Uses)
        (polypropylene fibers pigmented by, light stabilizers for, hindered
        amine-metal phosphonate mixts. as)
ΙT
     20437-10-9, C.I. Pigment Orange 65
     RL: USES (Uses)
        (polypropylene fibers pigmented by, light stabilizers for, hindered
        amine-metal phosphonate mixts. as)
RN
     20437-10-9 CAPLUS
CN
     Nickel, [[1,1'-[1,2-phenylenebis[(nitrilo-.kappa.N)methylidyne]]bis[2-
     naphthalenolato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
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PAGE 1-A

PAGE 2-A

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L12 ANSWER 12 OF 29 CAPLUS COPYRIGHT 1999 ACS AN 1993:126450 CAPLUS
```

DN 118:126450

ΤI Stabilization of polyamide fibers against heat and light by copper complex

compounds and oxalic acid diarylamides

Kaschig, Juergen; Reinert, Gerhard Ciba-Geigy A.-G., Switz. Eur. Pat. Appl., 30 pp. IN

PΑ

SO

CODEN: EPXXDW

DΤ Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND DATE	APPLICATION NO.	DATE
PI	EP 511166 R: AT, BE,	A1 19921028 CH, DE, DK, ES, FR,	EP 92-810286 GB. GR. IT. LI. LU	
	US 5338319 CA 2067059	A 19940816 AA 19921027	US 92-870650	19920420 19920424

Page 23

BR 9201524 19921201 BR 92-1524 19920424 Α JP 05186969 A2 19930727 JP 92-105104 19920424 PRAI CH 91-1252 19910426 MARPAT 118:126450 OS GI

Polyamide fibers are stabilized against heat and light by sulfo AB group-contg. derivs. of PhNHCOCONHPh and Cu complexes of o-azomethine phenol derivs. showing good affinity for polyamides. Polyamide fibers dyed with a mixt. of Cr complex azo dyes in the presence of I and II showed better lightfastness than similar fibers contg. no I.

II

IT 14167-15-8P

> RL: IMF (Industrial manufacture); PREP (Preparation) (prepn. of, as antioxidant for polyamide fibers)

IC ICM D06P001-649

> TCS D06P003-24; D06P001-642

CC 40-9 (Textiles and Fibers)

Section cross-reference(s): 25

ST polyamide fiber heat light stabilizer; copper complex antioxidant polyamide fiber; amide antioxidant polyamide fiber; sulfonate oxamide deriv stabilizer polyamide

IT

Amides, preparation RL: IMF (Industrial manufacture); PREP (Preparation) (prepn. and antioxidant activity in polyamide fibers)

145233-75-6P 145233-77-8P 145233-78-9P 145233-79-0P IT 145233-80-3P 145233-81-4P

RL: IMF (Industrial manufacture); PREP (Preparation)

(prepn. and antioxidant activity in polyamide fibers)

ΙT 14167-15-8P 145233-82-5P 145233-84-7P 145233-83-6P 145233-89-2P 145233-85-8P 145233-86-9P 145233-87-0P 145233-88-1P 145233-90-5P 145233-91-6P 145233-94-9P 145233-92-7P 145233-93-8P 145233-98-3P 145233-95-0P 145233-96-1P 145233-99-4P 145233-97-2P

145234-01-1P 145234-00-0P 145234-02-2P 145234-03-3P

RL: IMF (Industrial manufacture); PREP (Preparation) (prepn. of, as antioxidant for polyamide fibers)

IT 14167-15-8P

> RL: IMF (Industrial manufacture); PREP (Preparation) (prepn. of, as antioxidant for polyamide fibers)

RN 14167-15-8 CAPLUS CN Copper,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

```
L12 ANSWER 13 OF 29 CAPLUS COPYRIGHT 1999 ACS
```

AN 1990:531727 CAPLUS

DN 113:131727

TI Preparation of biphenyldiols as materials for resins

IN Kitamura, Taku; Kurokawa, Noriko

PA Dainippon Ink and Chemicals, Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PΙ

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01283241	A2	19891114	JP 88-111553	19880510

OS MARPAT 113:131727

- AB Biphenyldiols, also useful as antioxidants (no data) and materials for drugs and agrochems., are prepd. by liq.-phase oxidn. of molten phenols in presence of metal complex catalysts and successive heating without O. Thus, 103.0 g molten 2,6-di-tert-butylphenol (I) was heated with Mn phthalocyanine at 160.degree. under 140 mL/min air for 8 h to give a reaction mixt. contg. 3,3',5,5'-tetra-tert-butyl-4,4'-dihydroxybiphenyl (II) 25.6, 3,3',5,5'-tetra-tert-butyl-4,4'-diphenoquinone (III) 9.9, and I 62.3%, which was treated continuously without O at 180.degree. for 6 h to give a mixt. contg. II 45.0, III 0.2, and I 52.6%, from which 40.1 g cryst. II was isolated.
- IT 14167-18-1 23755-16-0

RL: CAT (Catalyst use); USES (Uses)

(catalyst, for liq.-phase oxidative coupling of phenols)

IC ICM C07C039-15

ICS B01J031-22; C07C037-11

ICA C07B061-00

CC 25-10 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds) Section cross-reference(s): 1, 35, 41

ST biphenyldiol prepn resin material; antioxidant biphenyldiol prepn; drug intermediate biphenyldiol prepn; dye intermediate biphenyldiol

prepn; phenol liq phase oxidative coupling; disproportionation phenol diphenoquinone ΙT Antioxidants (biphenyldiols) 132-16-1, Iron phthalocyanine 13930-88-6 14167-18-1 ΙT 14325-24-7, Manganese phthalocyanine 15665-27-7 17632-19-8 23755-16-0 RL: CAT (Catalyst use); USES (Uses) (catalyst, for liq.-phase oxidative coupling of phenols) 14167-18-1 23755-16-0 ΙT RL: CAT (Catalyst use); USES (Uses) (catalyst, for liq.-phase oxidative coupling of phenols) RN 14167-18-1 CAPLUS Cobalt, CN

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

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L12 ANSWER 14 OF 29 CAPLUS COPYRIGHT 1999 ACS
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AN 1988:456513 CAPLUS

DN 109:56513

TI Process for the photochemical stabilization of a fibrous polyamide material and its mixtures with other fibers

IN Reinert, Gerhard

PA Ciba-Geigy A.-G., Switz.

SO Eur. Pat. Appl., 28 pp.

CODEN: EPXXDW

DT Patent

LA German

FAN.CNT 1

CAN.	CNI	1						
	PA?	TENT NO.		KIND	DATE		APPLICATION NO.	DATE
ΡI	EP	255481		A1	19880203		EP 87-810415	19870723
		R: AT,	BE,	CH, DE	, ES, FR,	GB,	IT, LI, NL, SE	
	US	4874391		A	19891017		US 87-75805	19870720
	DK	8703934		Α	19880130		DK 87-3934	19870728
	AU	8776182		A1	19880204		AU 87-76182	19870728
	ΑU	604730		B2	19910103			
	BR	8703897		Α	19880405		BR 87-3897	19870728
	JΡ	63046262		A2	19880227		JP 87-187883	19870729
	JΡ	03064554		B4	19911007			
PRAT	CH	86-3034		19860	729			

PRAI CH 86-3034 19860

OS MARPAT 109:56513

The lightfastness and tear strength of polyamide fibers are increased by treating them with mixts. of H2O-sol. Cu complex dyes, light stabilizers, and, optionally, antioxidants. A nylon 66 fabric dyed in a bath contg. 1 g/L (NH4)2SO4, 0.043% 1:2 Cr-azo dye complex, and 0.063% 1:1 Cu azo dye complex had lightfastness (DIN 75.202) 6 and tenacity and elongation after lightfastness testing (150 h) 30.5 and 45.6%, resp., vs. 6, 9.0, and 14.7, resp., when dyed with Co complex dyes only.

IT 70882-93-8 115402-47-6

RL: USES (Uses)

(light stabilizers, for polyamide fibers in dyeing)

IC ICM D06P003-24 D06P001-62; D06P001-64; D06P001-653; D06P001-667; D06M013-10; ICS D06M013-26; D06M013-34 40-6 (Textiles and Fibers) CC Section cross-reference(s): 41 117-99-7D, 2-Hydroxybenzophenone, derivs. 14363-26-9 20170-32-5D, ΙT esters with polyols 52829-07-9, Bis(2,2,6,6-tetramethyl-4piperidinyl)sebacate 59261-49-3 66165-37-5 **70882-93-8** 101196-79-6 92484-54-3 103193-42-6 115402-47-6 RL: USES (Uses) (light stabilizers, for polyamide fibers in dyeing) 70882-93-8 115402-47-6 ΙT RL: USES (Uses) (light stabilizers, for polyamide fibers in dyeing) 70882-93-8 CAPLUS RN Cuprate(2-), [[3,3'-[1,2-ethanediylbis(nitrilomethylidyne)]bis[4-CN hydroxybenzenesulfonato]](4-)-N3,N3',O4,O4']-, disodium, (SP-4-2)- (9CI) (CA INDEX NAME)

• 2 Na+

RN 115402-47-6 CAPLUS
CN Cuprate(2-), [[3,3'-[1,2-ethanediylbis(nitrilomethylidyne)]bis[4,6-dihydroxybenzenesulfonato]](4-)-N3,N3',O4,O4']-, disodium (9CI) (CA INDEX NAME)

2 Na+

L12 ANSWER 15 OF 29 CAPLUS COPYRIGHT 1999 ACS

1988:133369 CAPLUS AN

108:133369 DN

ΤI Process for the photochemical stabilization of undyed and dyed fibrous polyamide material and its mixture with other fibers

Reinert, Gerhard; Burdeska, Kurt IN

Ciba-Geigy A.-G. , Switz. PΑ

SO Eur. Pat. Appl., 49 pp.

CODEN: EPXXDW

DT Patent

LA German

FAN.CNT 1								
	PAT	TENT NO.	KIND	DATE		APPLICATION NO.	DATE	
ΡI	EΡ	245204	A1	19871111		EP 87-810272	19870429	
	EP	245204	B1	19930804				
		R: AT, B	E, CH, D	E, ES, FR, G	В,	IT, LI, SE		
	US	4775386	A	19881004		US 87-42771	19870427	
	ΑT	92552	E	19930815		AT 87-810272	19870429	
	ES	2058136	Т3	19941101		ES 87-810272	19870429	
	ΑU	8772472	A1	19871112		AU 87-72472	19870504	
	ΑU	599649	B2	19900726				
	ZA	8703171	A	19871230		ZA 87-3171	19870504	
	BR	8702227	Α	19880217		BR 87-2227	19870504	
	JΡ	62267367	A2	19871120		JP 87-110405	19870506	
	JΡ	04011589	B4	19920228				
PRAI	CH	86-1826	1986	0505				
	CH	86-5057	1986					
	EΡ	87-810272	1987	0429				

AΒ In the title process, giving improved fastness and tenacity, the fibers are treated with mixts. of org. Cu complexes, light stabilizers, and optionally antioxidants. Nylon 66 staple yarn dyed olive-green in a dyebath contg. 0.04% (based on yarn) 1:1 Cu complex with the 2:1

```
imine of salicylaldehyde with ethylenediamine and 1%
2-benzotriazol-2-yl-4-
    methylphenol had Xenotest lightfastness (Swiss std. SN-ISO 105-B02) 7 and
    tenacity retention 71.2 and elongation retention 61.3% after 750 h
    Xenotest exposure; vs. 7, 64.5, and 56.7, resp., without the chelate,
6-7,
     69.4, and 64.2, resp., without the phenol, and 6-7, 49.7, and 51.2,
resp.,
    with neither.
IT
    13928-30-8 14167-15-8 70882-93-8
    RL: USES (Uses)
        (light stabilizers, for dyed and undyed polyamide fibers)
IC
    ICM D06P003-24
         D06P001-62; D06P001-64; D06P001-667; D06P005-02; D06M013-50;
     ICS
          D06M013-34; D06M013-28; D06M013-10
CC
     40-9 (Textiles and Fibers)
IT
     2440-22-4
                 3121-60-6 13928-30-8 14167-15-8
                               52829-07-9
     14363-26-9
                  23128-74-7
                                            57877-92-6
                                                          59261-49-3
                  92484-54-3
                               101196-77-4
     70882-93-8
                                             101196-79-6
                                                            103193-42-6
     113644-20-5
    RL: USES (Uses)
        (light stabilizers, for dyed and undyed polyamide fibers)
IT
    13928-30-8 14167-15-8 70882-93-8
    RL: USES (Uses)
        (light stabilizers, for dyed and undyed polyamide fibers)
RN
    13928-30-8 CAPLUS
    Copper, [[4,4'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[1,3-
CN
    benzenediolato-.kappa.O3]](2-)]- (9CI) (CA INDEX NAME)
```

14167-15-8 CAPLUS

RN

CN Copper,
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 70882-93-8 CAPLUS

CN Cuprate(2-), [[3,3'-[1,2-ethanediylbis(nitrilomethylidyne)]bis[4-hydroxybenzenesulfonato]](4-)-N3,N3',O4,O4']-, disodium, (SP-4-2)- (9CI) (CA INDEX NAME)

• 2 Na+

L12 ANSWER 16 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1987:83273 CAPLUS

DN 106:83273

TI Antioxidants for foods in refrigerators.

IN Fujita, Yuko

PA Japan Storage Battery Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 3 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.

KIND DATE

APPLICATION NO. DATE

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                      A2
                            19860917
                                           JP 85-49747
                                                            19850312
    JP 61209041
PΙ
    Bis(salicylaldehydo)ethylenediimine cobalt (II) or its deriv. is placed
AB
    and sealed in a porous polytetrafluoroethylene bag for use as an
     antioxidant for foods in refrigerators. Thus, 2 g granules
    bis(salicylaldehydo)ethylenediamine cobalt (II) was sealed in a
    polytetrafluoroethylene bag (40% porosity, 0.1 mm thick) to obtain an
     antioxidant for the foods. For regeneration, the used bag was
     soaked in 60-100.degree. water or heated with a high frequency-induced
     elec. heater to remove O.
IT
     14167-18-1
     RL: BIOL (Biological study)
        (antioxidant, polytetrafluoroethylene bags contq., for foods
        in refrigeration)
IC
     ICM B01J020-34
     ICS B01D053-14
CC
     17-6 (Food and Feed Chemistry)
     antioxidant bissalicylaldehydoethylenediimine cobalt food
ST
IT
        (antioxidants for refrigerated, bis(salicylaldehydo)ethylened
        iimine cobalt in polytetrafluoroethylene bags as)
ΙT
     Refrigerating apparatus
        (bis(salicylaldehydo)ethylenediimine cobalt in polytetrafluoroethylene
        bags as antioxidant for foods in)
ΙT
    Antioxidants
        (for foods, bis(salicylaldehydo)ethylenediimine cobalt in
        polytetrafluoroethylene bags as)
ΙT
     14167-18-1
     RL: BIOL (Biological study)
        (antioxidant, polytetrafluoroethylene bags contg., for foods
        in refrigeration)
     9002-84-0
IT
     RL: BIOL (Biological study)
        (bags, bis(salicylaldehydo)ethylenediimine cobalt in, as
     antioxidant for food in refrigeration)
IT
     14167-18-1
     RL: BIOL (Biological study)
        (antioxidant, polytetrafluoroethylene bags contg., for foods
        in refrigeration)
     14167-18-1 CAPLUS
RN
    Cobalt,
CN
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
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L12 ANSWER 17 OF 29 CAPLUS COPYRIGHT 1999 ACS
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AN 1987:83108 CAPLUS

DN 106:83108

TI Role of alpha-tocopherol, ascorbic acid, citric acid and EDTA as oxidants in model systems

AU Mahoney, John R., Jr.; Graf, Ernst

CS Dep. Surg., Univ. Minnesota, Minneapolis, MN, 55455, USA

SO J. Food Sci. (1986), 51(5), 1293-6 CODEN: JFDSAZ; ISSN: 0022-1147

DT Journal

LA English

AB The effects of 4 widely employed antioxidants on Fe mediated hydroxyl radical formation and lipid peroxidn. were studied in aq. model systems. Fe and Cu served as catalysts for the reactions that oxidized ascorbic acid [50-81-7] and .alpha.-tocopherol [59-02-9] and reduced O. Fe2+ spontaneously reduced O to O2- (superoxide anion radical) which led to .bul.OH (hydroxyl radical) and H2O2 generation and lipid peroxidn. Pptn. or sequestration of Fe greatly depressed these oxidative events. Complexation by EDTA [60-00-4] and citric acid [77-92-9], however, formed catalytically active Fe chelates. The concomitant increase in Fe soly. explained the substantial enhancement of Fe-driven redox reactions by EDTA and citric acid.

IT 19441-99-7

RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study)

(oxidn. catalyst activity of, for ascorbic acid and lipids in food $\ensuremath{^\circ}$ models)

CC 17-2 (Food and Feed Chemistry)

ST prooxidant lipid copper; antioxidant iron lipid peroxidn; tocopherol iron lipid oxidn; ascorbate iron lipid oxidn; EDTA iron lipid oxidn; citrate iron lipid oxidn

IT Food

(antioxidants for, prooxidative effects of, in models)

IT Antioxidants

(prooxidative effects of, in food models)

IT 3269-25-8 7439-89-6D, dildhydroxybenzoate complexes 12706-08-0, Iron (III) EGTA 14836-73-8 15275-07-7, Iron (III) EDTA 16448-54-7 19441-99-7 20438-93-1 23383-11-1 23567-85-3 27138-57-4D, 30492-15-0 iron complexes 47379-04-4 51595-41-6 RL: BAC (Biological activity or effector, except adverse); BIOL

(Biological study)

(oxidn. catalyst activity of, for ascorbic acid and lipids in food models)

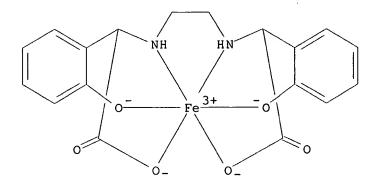
IT 19441-99-7

RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study)

(oxidn. catalyst activity of, for ascorbic acid and lipids in food models)

RN 19441-99-7 CAPLUS

CN Ferrate(1-), [[.alpha.,.alpha.'-(1,2-ethanediyldiimino)bis[2-hydroxybenzeneacetato]](4-)]-, hydrogen (9CI) (CA INDEX NAME)



● H+

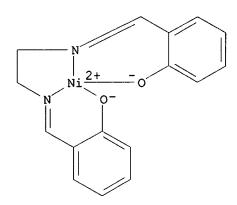
- L12 ANSWER 18 OF 29 CAPLUS COPYRIGHT 1999 ACS
- AN 1986:496860 CAPLUS
- DN 105:96860
- TI Catalysis of chain termination of oxidation of dialkyl sulfides by transition metal complexes
- AU Kovtun, G. A.; Lysenko, D. L.; Larin, G. M.; Rotov, A. V.; Moiseev, I. I.
- CS USSR
- SO Dokl. Akad. Nauk SSSR (1986), 287(6), 1418-21 [Phys. Chem.] CODEN: DANKAS; ISSN: 0002-3264
- DT Journal
- LA Russian
- AB Bis(salicylideneanilinato)copper (I) and bis(salicylidene-p-anisidinato)cobalt inhibited Pr2S oxidn. by catalyzing chain termination. Both the oxidized and reduced forms of Cu participated in the inhibition process in the case of I.
- IT 14167-20-5
 - RL: PRP (Properties)

(attempted inhibition by, of Pr sulfide oxidn.)

- CC 22-7 (Physical Organic Chemistry)
- ST antioxidant propyl sulfide metal complex; copper complex antioxidant propyl sulfide; cobalt complex antioxidant propyl sulfide
- IT Antioxidants

(transition metal complexes, for Pr sulfide)

14363-27-0 ΙT 14566-16-6 RL: PRP (Properties) (antioxidant, for Pr sulfide) 122-39-4, uses and miscellaneous 732-26-3 ΙT RL: USES (Uses) (antioxidant, for Pr sulfide) 14167-20-5 ΙT RL: PRP (Properties) (attempted inhibition by, of Pr sulfide oxidn.) ΙT 14167-20-5 RL: PRP (Properties) (attempted inhibition by, of Pr sulfide oxidn.) RN 14167-20-5 CAPLUS CN Nickel, [[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)



GI

L12 ANSWER 19 OF 29 CAPLUS COPYRIGHT 1999 ACS AN 1982:180465 CAPLUS DN 96:180465 ΤI Mechanism of inhibition of oxidation reactions by metal complexes. 7. Chain termination by copper complexes with the participation of alkyl and peroxyl radicals ΑU Kovtun, G. A.; Moiseev, I. I. Inst. Obshch. Neorg. Khim., Moscow, USSR
Izv. Akad. Nauk SSSR, Ser. Khim. (1982), (2), 260-5 CS so CODEN: IASKA6; ISSN: 0002-3353 DT Journal LA Russian

AB Kinetic data indicated that chain termination in the oxidm, of pentaerythritol esters of C5-C9 monocarboxylic acids in the presence of I,

II, and III involved reactions of alkyl and peroxy radicals with the Cu complexes.

IT 81565-93-7

RL: PRP (Properties)

(antioxidant, for pentaerythritol esters, mechanism of inhibition by)

CC 22-7 (Physical Organic Chemistry)

ST pentaerythritol ester antioxidant copper complex; alkyl radical reaction copper complex; peroxy radical reaction copper complex

IT Antioxidants

(copper complexes, for pentaerythritol esters, mechanism of inhibition by)

IT 78885-13-9 81565-92-6 **81565-93-7**

RL: PRP (Properties)

(antioxidant, for pentaerythritol esters, mechanism of inhibition by)

IT 81565-93-7

RL: PRP (Properties)

(antioxidant, for pentaerythritol esters, mechanism of inhibition by)

RN 81565-93-7 CAPLUS

CN Copper, [[2,2'-[1,2-ethanediylbis(nitrilomethylidyne)]bis[4-(1,1-dimethylethyl)phenolato]](2-)-N,N',O,O']- (9CI) (CA INDEX NAME)

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ANSWER 20 OF 29 CAPLUS COPYRIGHT 1999 ACS
L12
AN
     1981:409750 CAPLUS
DN
     95:9750
TΙ
     Oil for use in automobile engines and transmissions
     Gimpirea, Marin; Herdan, Jean Michel
IN
     Institutul de Cercetari si Proiectari Tehnologice pentru Rafinarii si
PA
     Instalatii Petrochimice, Rom.
     Rom., 3 pp.
SO
     CODEN: RUXXA3
DT
     Patent
     Romanian
T.A
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                            APPLICATION NO.
                                                              DATE
                                            -----
                             -----
                             19790616
                                       · RO 75-83619
PΙ
     RO 67396
                       В
                                                              19751016
     Lubricating oils for automobile engines and mech. and hydraulic
AB
     transmissions comprise SAE 20 base oil (contg. paraffins, naphthenes and
     naphthenes-paraffins), polymethacrylate (I) pour point depressant and
     viscosity index improver 1-4, phenyl-.beta.-naphthylamine (II)
[135-88-6]
     antioxidant 0.1-0.5, Zn dialkyldithiocarbamate (III) antiwear
     additive 0.1-2.5, overbased Ca sulfonate (IV) dispersant-detergent 2-6,
s-
     and P-contg. extreme-pressure additive 1.5-5, ash-free succinimide (V)
     dispersant-detergent 1-4, Cu 1,2-disalicylpropylene diamine (VI) [
     14522-52-2] corrosion inhibitor 0.01-0.1, and silicone antifoam
     additive 0.002%. Thus, lubricating oil (d420 0.900, viscosity at
     50.degree. 52 cSt, viscosity index 115, flash point 220.degree., contg.
Ca
     0.30, S 0.98, P 0.017, Zn 0.1, and N 0.087%) was obtained by blending
87.6
     kg SAE 20 base oil with I 3, II 0.4, III 1, IV 4, extreme-pressure additive contg. 25-30% S and 1-2% P 3, V 1, VI corrosion inhibitor 0.03,
     and silicone antifoam 0.002 kg for 0.5-1 h at 60-80.degree..
IT
     14522-52-2
     RL: USES (Uses)
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(lubricating oils and transmission fluids contq., for automobiles)

Page 37

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IC
     C10M001-001
     51-7 (Fossil Fuels, Derivatives, and Related Products)
CC
     79-41-4D, esters, polymers
                                  135-88-6
                                             594-07-0D, dialkyl esters, zinc
ΙT
     salts 14522-52-2
     RL: USES (Uses)
        (lubricating oils and transmission fluids contg., for automobiles)
ΙT
     14522-52-2
     RL: USES (Uses)
        (lubricating oils and transmission fluids contg., for automobiles)
RN
     14522-52-2 CAPLUS
     Copper, [[2,2'-[[(1R)-1-methyl-1,2-ethanediyl]bis[(nitrilo-
CN
     .kappa.N)methylidyne]]bis[phenolato-.kappa.O]](2-)]-, (SP-4-4)- (9CI)
(CA
     INDEX NAME)
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ANSWER 21 OF 29 CAPLUS COPYRIGHT 1999 ACS

L12

```
AN
     1981:158081 CAPLUS
DN
     94:158081
TI
    Light stabilization of thermoplastic elastomers
     Kovshov, Yu. S.; Moiseev, V. V.; Zharkikh, T. P.; Safonova, V. P.
ΑU
CS
     Voronezh. Filial, Vses. Nauchno-Issled. Inst. Sint. Kauch., Voronezh,
USSR
SO
    Prom-st. Sint. Kauch. (1980), (12), 17-20
    CODEN: PSKAD6
DT
     Journal
LA
    Russian
AB
    Of the 19 antioxidants and light stabilizers tested with DST-30
     and DMST-30 thermoplastic elastomers, 2-(2-hydroxy-5-
    methylphenyl)benzotriazole (Tinuvin P) (I) [2440-22-4],
    N, N'-bis(2, 2, 6, 6-tetramethyl-4-piperidyl)ethylenediamine [61260-54-6],
    Tinuvin 328 [25973-55-1], and Ni-contg. compds., e.g., nickel stearate
     [2223-95-2], nickel diethyldithiocarbamate [14267-17-5],
    N, N'-bis(2-hydroxy-4-methoxyphenylbenzylidene)ethylenediamine nickel
    complex [77110-56-6], diphenylsilanediol nickel salt
     [77077-11-3], and oxalic acid p-methoxyanilide nickel salt [77077-12-4].
    The effectiveness of I can be increased nearly 2-fold by addn. of EV-1
    wax. In all cases, the Ni-contg. compds. were more effective than
     2,4-dihydroxybenzophenone. Although the stabilizers increased the light
    resistance of the thermoplastic elastomers, the abs. light resistance of
     the stabilized thermoplastic elastomers remained quite low.
```

IT 77110-56-6

RL: USES (Uses)

(light stabilizers, for thermoplastic elastomers)

CC 38-9 (Elastomers, Including Natural Rubber)

IT 2223-95-2 2440-22-4 14267-17-5 25973-55-1 61260-54-6 77077-11-3

77077-12-4 **77110-56-6**

RL: USES (Uses)

(light stabilizers, for thermoplastic elastomers)

IT 77110-56-6

RL: USES (Uses)

(light stabilizers, for thermoplastic elastomers)

RN 77110-56-6 CAPLUS

CN Nickel, [[4,4''-[1,2-ethanediylbis(nitrilomethylidyne)]bis[ar'-methoxy[1,1'-biphenyl]-3-olato]](2-)-N4,N4'',O3,O3'']- (9CI) (CA INDEX NAME)

PAGE 1-A

2 (D1-O-Me)

PAGE 2-A

L12 ANSWER 22 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1979:186069 CAPLUS

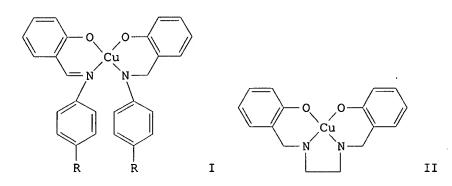
DN 90:186069

TI Mechanism of the inhibition of oxidation reactions by metal complexes.

 Effect of substituents in the imino component of a Schiff base on the antioxidant properties of copper salicylaldiminates

AU Kovtun, G. A.; Lysenko, D. L.; Berenblyum, A. S.; Moiseev, I. I.

CS Inst. Obshch. Neorg. Khim. im. Kurnakova, Moscow, USSR SO Izv. Akad. Nauk SSSR, Ser. Khim. (1979), (2), 293-7 CODEN: IASKA6; ISSN: 0002-3353
DT Journal Russian
GI



AB I (R = H, OH, Me, MeO, NO2) and II participated repeatedly in chain terminations (in the oxidn. of cyclohexylamine) via reaction with .alpha.-amino peroxy radicals (III). The rate consts. of single-electron oxidn. and redn. of III decreased with increasing donor character of R. The rates of chain termination by (4-RC6H4)2NH increased with increasing donor character of R.

IT 14167-15-8

RL: USES (Uses)

(inhibition of cyclohexylamine oxidn by, parameters of)

CC 22-5 (Physical Organic Chemistry)

ST copper complex **antioxidant** cyclohexylamine; cyclohexylamine oxidn kinetics inhibition

IT 122-39-4, uses and miscellaneous 620-93-9 1821-27-8 6962-04-5 14167-15-8 14363-27-0 14688-83-6 14852-50-7 14852-76-7 15412-33-6

RL: USES (Uses)

(inhibition of cyclohexylamine oxidn by, parameters of)

IT 14167-15-8

RL: USES (Uses)

(inhibition of cyclohexylamine oxidn by, parameters of)

RN 14167-15-8 CAPLUS

CN Copper,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

L12 ANSWER 23 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1979:123051 CAPLUS

DN 90:123051

TI Bis(azomethine) pigments

IN Mowat, Douglas

PA Ciba-Geigy A.-G., Switz.

SO Ger. Offen., 21 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

ran.cni i						
		PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	ΡI	DE 2823438	A1	19781214	DE 78-2823438	19780529
		GB 1564231	Α	19800402	GB 77-22899	19770531
		US 4198345	Α	19800415	US 78-908182	19780522
		CA 1111436	A1	19811027	CA 78-304333	19780529
		CH 637418	Α	19830729	CH 78-5841	19780529
		JP 53149226	A2	19781226	JP 78-65625	19780531
		FR 2393033	A1	19781229	FR 78-16214	19780531
		FR 2393033	В1	19800606		
	PRAI	GB 77-22899	19770	531		
	GI					

Ι

AB Azomethine pigment I [20437-10-9], used for coloring coatings and polyethylene [9002-88-4] fast orange-red shades, is prepd. by reaction of 2-hydroxy-1-naphthaldehyde (II) [708-06-5] 5.45 with o-phenylenediamine [95-54-5] 3.425 in the presence of antioxidant Na2S2O5 5 kg and a nonionic surfactant in H2O at 20-5.degree., heating to Page 41

90.degree., adding nickel nitrate and NH4OH, heating, adding 5.5 kg II, heating at 95-100.degree., and isolating.

IT 20437-10-9

RL: USES (Uses)

(pigment, for coatings and polyethylene, prepn. of)

IC C09B055-00

CC 40-4 (Dyes, Fluorescent Whitening Agents, and Photosensitizers)

ST azomethine nickel hydroxynaphthaldehyde pigment; phenylenediamine azomethine nickel pigment; antioxidant nickel azomethine pigment

IT 7681-57-4

RL: USES (Uses)

(antioxidants, in prepn. of bisazomethine pigments)

IT 20437-10-9

RL: USES (Uses)

(pigment, for coatings and polyethylene, prepn. of)

IT 20437-10-9

RL: USES (Uses)

(pigment, for coatings and polyethylene, prepn. of)

RN 20437-10-9 CAPLUS

CN Nickel, [[1,1'-[1,2-phenylenebis[(nitrilo-.kappa.N)methylidyne]]bis[2-naphthalenolato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 2-A



```
L12 ANSWER 24 OF 29 CAPLUS COPYRIGHT 1999 ACS
AN
     1977:405019 CAPLUS
DN
     87:5019
     Inhibition of oxidative induced decomposition of (.alpha..alpha.'-
ΤI
     diphenyl)azoethane by transition metal ions. Interactions of metal ions
     with .alpha.-phenylethylperoxyl radical
     Okuma, Kazuhiko; Niki, Etsuo; Kamiya, Yoshio
ΑU
     Dep. React. Chem., Univ. Tokyo, Tokyo, Japan
CS
     J. Chem. Soc., Perkin Trans. 2 (1977), (1), 59-64
SO
     CODEN: JCPKBH
     Journal
DT
     English
LA
     MeCHPhN:NCHPhMe (I) decompd. >102 times as fast in the presence of O than
AB
     in its absence; this oxidative-induced decompn. was suppressed by radical
     scavengers and transition metal ions in both lower and higher valence
     states. The effect of the transition metal ions was ascribed to their
     fast interactions with peroxyl radicals and their stabilization of the
     latter. At lower metal concns. the induced decompn. was obsd. after a
     specific suppression time detd. by the ratio of concns. of I to metal
ion.
     The effect of ligand and additives such as pyridine and carboxylic acids
     on the rate and products of the decompn. is discussed.
ΙT
     14167-18-1
     RL: USES (Uses)
        (inhibitor, for oxidative decompn. of diphenylazoethane)
CC
     22-4 (Physical Organic Chemistry)
IT
     Antioxidants
        (transition metal ions, for decompn. of diphenylazoethane)
IT
     7436-86-4
                 14024-48-7
                              14024-58-9
                                           14033-48-8 14167-18-1
     14284-89-0
                  14325-24-7
     RL: USES (Uses)
        (inhibitor, for oxidative decompn. of diphenylazoethane)
     14167-18-1
IT
     RL: USES (Uses)
        (inhibitor, for oxidative decompn. of diphenylazoethane)
RN
     14167-18-1 CAPLUS
CN
     Cobalt,
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
```

```
L12
    ANSWER 25 OF 29 CAPLUS COPYRIGHT 1999 ACS
     1976:524904 CAPLUS
AN
     85:124904
DN
     Mechanisms of inhibition against the copper-catalyzed oxidation of
ΤI
     polyethylene: structures and catalytic reactivities of copper-inhibitor
     complexes
ΑU
     Allara, D. L.; Chan, M. G.
     Bell Lab., Murray Hill, N. J., USA
CS
     J. Polym. Sci., Polym. Chem. Ed. (1976), 14(8), 1857-76
SO
     CODEN: JPLCAT
DT
     Journal
LA
     English
     The oxidn. rates were detd. for polyethylene [9002-88-4] contg. various
AΒ
     copper complexes presumed to form in situ in polyethylene contg. copper
     salts as inhibitors of copper-catalyzed oxidn. The structure-catalytic
     activity correlations were studied. The complexes with neg. charged {\tt N}
     ligands gave low reactivities; the highest reactivities were shown by
Cu<sub>2</sub>0
     and air-exposed metallic Cu. The mechanism and kinetics of the
inhibition
     are discussed.
IT
     14167-15-8
     RL: USES (Uses)
        (prepn. of and polyethylene oxidn. kinetics in presence of)
CC
     36-6 (Plastics Manufacture and Processing)
IT
     Antioxidants
        (oxamides and other copper-complexing compds., for polyethylene in
        contact with copper)
                  27721-61-5
IT
     14167-15-8
                                60650-17-1
                                             60749-19-1
                                                           60801-62-9
     60953-61-9
                  60955-16-0
                                60955-17-1
                                             60955-18-2
                                                           60955-19-3
     60955-20-6
     RL: USES (Uses)
        (prepn. of and polyethylene oxidn. kinetics in presence of)
IT
     14167-15-8
     RL: USES (Uses)
        (prepn. of and polyethylene oxidn. kinetics in presence of)
RN
     14167-15-8 CAPLUS
CN
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
```

```
L12 ANSWER 26 OF 29 CAPLUS COPYRIGHT 1999 ACS
ΑN
     1976:136642 CAPLUS
     84:136642
DN
     Stabilization of acrylonitrile-styrene-butadiene polymers
TI
     Sziburies, Ute; Schlimpes, Rolf; Schroeder, Elisabeth
IN
PA
     E. Ger.
SO
     Ger. (East), 3 pp.
     CODEN: GEXXA8
DT
     Patent
LA
     German
FAN.CNT 1
                                           APPLICATION NO.
     PATENT NO.
                      KIND DATE
                                                            DATE
                           _____
                                           _____
                            19750620
                                           DD 74-179985
                       Z
                                                            19740718
PΤ
     DD 113764
     For diagram(s), see printed CA Issue.
GT
AB
     The Ni complexes I (R1, R2 = alkyl, aryl or R1 + R2 = hydrocarbylene; R3,
     R4 = H, alkyl, aryl) are antioxidants and heat stabilizers for
     ABS polymer [9003-56-9]. Thus, ABS contg. 0.25% antioxidant KS
     and 0.20% I (R1 + R2 = o-C6H4, R3, R4 = H) [14406-71-4] has
     tensile strength 366, 354, 330, and 318 kg/cm2 after 0, 266, 463, and 803
     hr, resp., of exposure to sunlight; compared with 366, 246, 280, and 240,
     resp., with a conventional stabilizer.
     14167-20-5 14406-71-4
TΤ
     RL: USES (Uses)
        (antioxidants and heat stabilizers, for ABS)
     C08F; C08D
IC
CC
     36-6 (Plastics Manufacture and Processing)
ST
     ABS polymer stabilizer; heat stabilizer ABS; antioxidant ABS
     polymer; nickel complex stabilizer; salicylaldehyde imine complex nickel;
     phenylenediamine salicylidene complex nickel
IT
     Antioxidants
     Heat stabilizers
        (nickel salicylideneimine complexes, for ABS)
     9003-56-9
ΙT
     RL: USES (Uses)
        (antioxidants and heat stabilizers for, nickel complexes as)
     14167-20-5 14406-71-4
ΙT
     RL: USES (Uses)
        (antioxidants and heat stabilizers, for ABS)
```

IT 14167-20-5 14406-71-4

RL: USES (Uses)

(antioxidants and heat stabilizers, for ABS)

RN 14167-20-5 CAPLUS

CN Nickel,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 14406-71-4 CAPLUS

CN Nickel,

[[2,2'-[1,2-phenylenebis[(nitrilo-.kappa.N)methylidyne]]bis[phenol
ato-.kappa.O]](2-)]- (9CI) (CA INDEX NAME)

L12 ANSWER 27 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1975:532486 CAPLUS

DN 83:132486

TI Singlet oxygen and polymer photooxidations. II. Photodegradation of an olefinically unsaturated polymer

```
Zweig, A.; Henderson, W. A., Jr.
ΑU
    Chem. Res. Div., Am. Cyanamid Co., Stamford, Conn., USA
CS
     J. Polym. Sci., Polym. Chem. Ed. (1975), 13(4), 993-1014
SO
    CODEN: JPLCAT
DT
     Journal
LA
    English
    Ni amino(thiobisphenolate) type chelates were quant. more effective in
AB
     retarding XT 375 (acrylonitrile-1,3-butadiene-methyl methacrylate-styrene
     graft copolymer) [9010-94-0] photodegrdn. than were other additives with
    greater singlet O quenching efficiencies, uv absorption, or radical
     antioxidant properties. The Ni complexes were relatively stable
     under photoexposure conditions and were able to quench singlet O and
    precursor excited states. Fluoranthene [206-44-0] photosensitization of
    XT 375 was retarded by the Ni chelates.
IT
    14167-20-5 42532-94-5 55911-89-2
    RL: USES (Uses)
        (antioxidant light stabilizers, for acrylic graft polymers)
CC
     36-6 (Plastics Manufacture and Processing)
ST
    nickel chelate light stabilizer; antioxidant acrylic polymer;
    mechanism photodegrdn acrylic polymer
IT
    Antioxidants
        (for acrylic graft polymers)
    Ethanedithione, diphenyl-, nickel complexes
IT
     RL: USES (Uses)
        (antioxidant light stabilizers, for acrylic graft polymers)
IT
     Phosphorous acid, triphenyl ester, alkylated
    RL: USES (Uses)
        (antioxidants contg., for acrylic polymers)
                                                        14363-27-0
IT
     13927-77-0 14167-20-5 14266-60-5 14283-99-9
                  14568-00-4
                               14642-58-1
                                            15170-64-6
                                                          15412-63-2
     14516-71-3
                  15550-15-9
                               15711-57-6
    15415-93-7
                                            15748-52-4
                                                          16592-95-3
                  20649-88-1
                               21246-17-3
                                            28042-64-0
                                                          29666-10-2
     19587-52-1
                  37981-00-3
                               41754-10-3 42532-94-5
                                                        42950-64-1
     37025-91-5
                               55911-88-1 55911-89-2
                                                        55917-80-1
     42957-85-7
                  42957-88-0
     55917-83-4
                  55917-85-6
                               55948-34-0
                                            56724-26-6
                                                          56725-54-3
                               56725-57-6
                                            56725-58-7
    56725-55-4
                  56725-56-5
                                                          56725-59-8
    RL: USES (Uses)
        (antioxidant light stabilizers, for acrylic graft polymers)
IT
     56694-35-0
    RL: USES (Uses)
        (antioxidants contg., for acrylic polymers)
                          6683-19-8
IT
     119-47-1
                2082-79-3
                                        27676-62-6
    RL: USES (Uses)
        (antioxidants, for acrylic graft polymers)
TΤ
     14167-20-5 42532-94-5 55911-89-2
     RL: USES (Uses)
        (antioxidant light stabilizers, for acrylic graft polymers)
RN
     14167-20-5 CAPLUS
    Nickel,
CN
[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno
     lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)
```

RN 42532-94-5 CAPLUS

CN Nickel,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)ethylidyne]]bis[phenol ato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN

55911-89-2 CAPLUS Nickel, [[2,2'-[1,2-ethanediylbis(nitrilodecylidyne)]bis[4-methylphenolato]](2-)-N,N1,O,O1]- (9CI) (CA INDEX NAME) CN

IT

14167-18-1

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for autoxidn. of fatty acid esters)

```
L12 ANSWER 28 OF 29 CAPLUS COPYRIGHT 1999 ACS
     1973:83595 CAPLUS
ΔN
DN
     78:83595
     Autoxidation of fatty acid esters in the presence of a heavy metal
ΤI
     catalyst Salcomine (cobalt[II] bis(salicylalethylenediimine)). I.
Effect
     of catalyst upon rate of consumption of oxygen and decomposition of
     hydroperoxide
ΑU
     Jarvi, Pentti K.
CS
     Univ. Helsinki, Helsinki, Finland
SO
     Lipids (1972), 7(12), 755-61
     CODEN: LPDSAP
DT
     Journal
LA
     English
GI
     For diagram(s), see printed CA Issue.
     The autoxidn. of fatty acid esters in the presence of a heavy metal
     chelate, Salcomine (I) was studied. Both antioxidative and
     prooxidative effects were obsd. When the concn. of the catalyst is
     decreased or the temp. is increased, the induction period becomes shorter
     and under some conditions disappears. It was shown that the decompn. of
     hydroperoxides is affected by I; a first order reaction is involved with
     Arrhenius parameters E = 21 kcal, and A = 8.99 .times. 1010 sec-1. The
     solubility of O in fatty acid esters above 30.degree., with and without
I,
     is the same. Theoretical aspects of the reaction mechanism, that are
     consistent with the exptl. results, are proposed.
ΙT
     14167-18-1
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts, for autoxidn. of fatty acid esters)
     22-5 (Physical Organic Chemistry)
CC
     14167-18-1
ΙT
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts, for autoxidn. of fatty acid esters)
```

RN 14167-18-1 CAPLUS

CN Cobalt,

[[2,2'-[1,2-ethanediylbis[(nitrilo-.kappa.N)methylidyne]]bis[pheno lato-.kappa.O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

L12 ANSWER 29 OF 29 CAPLUS COPYRIGHT 1999 ACS

AN 1972:564319 CAPLUS

DN 77:164319

TI Stabilization of nitrosobenzene

IN Dodman, David; Wilkins, Malcolm; Woolley, John Mathers

PA Imperial Chemical Industries Ltd.

SO Ger. Offen., 6 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
PI DE 2207493	· A 19720907	DE 72-2207493	19720217
US 3751490	A 19730807	US 72-220703	19720125
PRAI GB 71-5095	19710222		

AB A 20 soln. of PhNO, useful as an intermediate in the manuf. of antioxidants and dyes, in PhNO2 was stabilized by 0.25-1 salts or complexes of Co, Cu, Mn, Hg, or Ce. Thus, a 20 soln. of PhNO in Ph-NO2 contg. 0.25 Mn acetate stored in the dark at 20-5.degree. for 7 and 34 days contained 96.2 and 90.8 of the initial PhNO, resp., vs. 81.6 and 39.6

without Mn acetate.

IT 36870-54-9

RL: RCT (Reactant)

(stabilizers, for nitrosobenzene)

IC CO7C

CC 26-6 (Condensed Aromatic Compounds)

Section cross-reference(s): 40

IT 638-38-0 1317-38-0, uses and miscellaneous 1600-27-7 7758-89-6 14284-89-0 19475-87-7 **36870-54-9** 38845-52-2

RL: RCT (Reactant)

(stabilizers, for nitrosobenzene)

IT 36870-54-9

RL: RCT (Reactant)

```
(stabilizers, for nitrosobenzene)
     36870-54-9 CAPLUS
RN
=> d .ca 113 1-7
L13 ANSWER 1 OF 7 CAPLUS COPYRIGHT 1999 ACS
     1998:705968 CAPLUS
AN
DN
     129:326110
TI
     Synthetic catalytic free radical scavengers useful as antioxidants
     for prevention and therapy of disease
IN
     Malfroy-Camine, Bernard; Doctrow, Susan Robin
PA
     Eukarion, Inc., USA
SO
     U.S., 51 pp. Cont.-in-part of U.S. 5,403,834.
     CODEN: USXXAM
DT
     Patent
LA
     English
FAN.CNT 5
                      KIND DATE
     PATENT NO.
                                           APPLICATION NO.
                                                             DATE
                            _____
     US 582<u>7880</u> 🛠
PΙ
                       Α
                            19981027
                                            US 95-380731
                                                             19950126
     US 5403834
                            19950404
                                           US 92-987474
                       Α
                                                             19921207
                                           CA 93-2150937
  —>CA 2150937
                       AA
                            19940623
                                                             19931206
                                           HU 95-1644
     HU 72967
                       A2
                            19960628
                                                             19931206
     GB 2305107
                                           GB 96-21087
                       Α1
                            19970402
                                                             19931206
     GB 2305107
                       В2
                            19970514
     LV 10924
                            19961020
                                           LV 95-158
                                                             19950606
                       В
                       Α
                            19971209
                                           US 95-485489
     US 5696109
                                                             19950607
                                           US 95-479697
     US 5834509
                       Α
                            19981110
                                                             19950607
PRAI US 92-987474
                      19921207
                      19931206
     WO 93-US11857
     GB 94-15050
                      19940706
     US 95-380731
                      19950126
     Antioxidant salen-metal complexes are provided. Salen-metal
AB
     complexes having superoxide activity, catalase activity, and/or
peroxidase
     activity, in a form suitable for pharmaceutical administration to treat
or
     prevent a disease assocd. with cell or tissue damage produced by free
     radicals such as superoxide, and cosmetic and free radical quenching
     formulations of salen metal compds. are also disclosed. The in vitro
     superoxide dismutase and catalase activities of the various salen-Mn
     complexes were tested. Formulations of topical lotions contg.
salen-metal
     complexes are presented.
     53140-26-4 53177-12-1 81065-76-1
     151434-18-3 186299-34-3 186358-92-9
     200485-53-6 200485-54-7 200485-55-8
     200577-45-3 200577-46-4 200577-48-6
     200577-49-7 200577-50-0 215112-77-9
     215112-79-1 215112-83-7 215112-84-8
     215112-85-9D, acylated
     RL: BAC (Biological activity or effector, except adverse); BUU
(Biological
     use, unclassified); THU (Therapeutic use); BIOL (Biological study); USES
     (Uses)
        (synthetic catalytic free radical scavengers useful as
```

antioxidants for prevention and therapy of disease)

```
IC
     ICM A61K031-28
         C07F013-00
     ICS
NCL
    514492000
CC
     1-12 (Pharmacology)
     Section cross-reference(s): 62, 63
ST
     synthetic radical scavenger antioxidant therapy disease
IT
     Injury
        (animal tissue; synthetic catalytic free radical scavengers useful as
      antioxidants for prevention and therapy of disease)
TΤ
     Radicals, biological studies
     RL: BSU (Biological study, unclassified); BIOL (Biological study)
        (free, scavengers; synthetic catalytic free radical scavengers useful
        as antioxidants for prevention and therapy of disease)
IT
    Antioxidants
     Lotions (drug delivery systems)
     Therapy
        (synthetic catalytic free radical scavengers useful as
      antioxidants for prevention and therapy of disease)
     53140-26-4 53177-12-1 81065-76-1
TΤ
     151434-18-3 186299-34-3 186358-92-9
     200485-53-6 200485-54-7 200485-55-8
     200577-45-3 200577-46-4 200577-48-6
     200577-49-7 200577-50-0 215112-77-9
     215112-79-1 215112-83-7 215112-84-8
     215112-85-9D, acylated
     RL: BAC (Biological activity or effector, except adverse); BUU
(Biological
     use, unclassified); THU (Therapeutic use); BIOL (Biological study); USES
     (Uses)
        (synthetic catalytic free radical scavengers useful as
      antioxidants for prevention and therapy of disease)
     9001-05-2, Catalase
                         9003-99-0, Peroxidase
                                                   9054-89-1, Superoxide
IT
     dismutase
     RL: BSU (Biological study, unclassified); BIOL (Biological study)
        (synthetic catalytic free radical scavengers useful as
      antioxidants for prevention and therapy of disease)
L13 ANSWER 2 OF 7 CAPLUS COPYRIGHT 1999 ACS
AN
     1998:53474 CAPLUS
DN
     128:200984
     Synthetic combined superoxide dismutase/catalase mimetics are protective
TΙ
     as a delayed treatment in a rat stroke model: a key role for reactive
     oxygen species in ischemic brain injury
ΑU
     Baker, Keith; Marcus, Catherine Bucay; Huffman, Karl; Kruk, Henry;
     Malfroy, Bernard; Doctrow, Susan R.
CS
     Eukarion, Inc., Bedford, MA, USA
     J. Pharmacol. Exp. Ther. (1998), 284(1), 215-221
SO
     CODEN: JPETAB; ISSN: 0022-3565
PB
     Williams & Wilkins
DT
     Journal
LA
     English
AB
     Stroke is a severe and prevalent syndrome for which there is a great need
     for treatment, including agents to block the cascade of brain injury that
     occurs in the hours after the onset of ischemia. Reactive oxygen species
     (ROS) have been implicated in this destructive process, but
     antioxidant enzymes such as superoxide dismutase (SOD) have been
     unsatisfactory in exptl. stroke models. This study is an evaluation of
                                                                        Page 52
```

the effectiveness of salen-manganese complexes, a class of synthetic SOD/catalase mimetics, in a rat focal ischemia model involving middle cerebral artery occlusion. The authors focus on EUK-134, a newly reported

salen-manganese complex demonstrated here to have greater catalase and cytoprotective activities and equiv. SOD activity compared with the previously described prototype EUK-8. The administration of EUK-134 at 3 h after middle cerebral artery occlusion significantly reduced brain infarct size, with the highest dose apparently preventing further infarct growth. EUK-8 was also protective but substantially less effective. These findings support a key role for ROS in the cascade of brain injury after stroke, even well after the onset of ischemia. The enhanced activity of EUK-134 suggests that, in particular, hydrogen peroxide contributes significantly to this injury. Overall, this study suggests that synthetic SOD/catalase mimetics might serve as novel,

multifunctional

therapeutic agents for stroke.

IT 81065-76-1P, EUK 134

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(synthetic combined superoxide dismutase/catalase mimetics are protective in delayed treatment in a rat stroke model and role for reactive oxygen species in ischemic brain injury)

IT 53177-12-1, EUK-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (synthetic combined superoxide dismutase/catalase mimetics are protective in delayed treatment in a rat stroke model and role for reactive oxygen species in ischemic brain injury)

CC 1-12 (Pharmacology)

Section cross-reference(s): 14

IT 81065-76-1P, EUK 134

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(synthetic combined superoxide dismutase/catalase mimetics are protective in delayed treatment in a rat stroke model and role for reactive oxygen species in ischemic brain injury)

IT **53177-12-1**, EUK-8

RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (synthetic combined superoxide dismutase/catalase mimetics are protective in delayed treatment in a rat stroke model and role for reactive oxygen species in ischemic brain injury)

- L13 ANSWER 3 OF 7 CAPLUS COPYRIGHT 1999 ACS
- AN 1997:809718 CAPLUS
- DN 128:80026
- TI Synthetic catalytic free radical scavengers useful as antioxidants for prevention and therapy of disease
- IN Malfroy-Camine, Bernard; Doctrow, Susan Robin
- PA Eukarion, Inc., USA
- SO U.S., 62 pp. Cont.-in-part of U.S. Ser. No. 380,731. CODEN: USXXAM
- DT Patent
- LA English

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FAN.CNT 5
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GI
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$$R^4$$
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AB The invention provides antioxidant salen-metal complexes,

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compns. of such antioxidant salen-metal complexes having
     superoxide activity, catalase activity, and/or peroxidase activity,
     compns. of salen-metal complexes in a form suitable for pharmaceutical
     administration to treat a disease assocd. with cell or tissue damage
     produced by free radicals such as superoxide, and cosmetic and free
     radical quenching formulations of salen metal compds. E.g., in vitro
     antioxidant catalytic activities and pharmaceutical formulations
    were given for salen metal compds. such as I.
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    Catalysts
     Drug delivery systems
    Radical scavengers
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    use); PRP (Properties); THU (Therapeutic use); BIOL (Biological study);
     USES (Uses)
        (salen manganese complexes as free radical scavenger
     antioxidant catalysts)
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1997:130022 CAPLUS
AN
DN
     126:135623
     Synthetic catalytic free radical scavengers useful as antioxidants
TТ
     for prevention and therapy of disease
     Malfroy-Camine, Bernard; Doctrow, Susan Robin
IN
     Eukarion, Inc., USA; Malfroy-Camine, Bernard; Doctrow, Susan Robin
PA
     PCT Int. Appl., 164 pp.
                                                                       JAM.
SO
     CODEN: PIXXD2
DT
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     English
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                            DATE
                                           APPLICATION NO.
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                      19950126
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                      19960606
OS
     MARPAT 126:135623
AB
     The invention provides antioxidant salen-metal complexes,
     compns. of such antioxidant salen-metal complexes having
     superoxide, catalase, and/or peroxidase activities, compns. of
salen-metal
     complexes in a form suitable for pharmaceutical administration to treat
or
     prevent a disease assocd. with cell or tissue damage produced by free
     radicals such as superoxide, and cosmetic and free radical quenching
     formulations of salen metal compds.
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     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
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     ICM A61K031-555
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     ICS A61K031-28; A61K031-295
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     63-6 (Pharmaceuticals)
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TΤ
     Drug delivery systems
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        (salen metal complexes as radical scavengers as antioxidants
        for prevention and therapy of disease)
L13 ANSWER 5 OF 7 CAPLUS COPYRIGHT 1999 ACS
     1997:121374 CAPLUS
ΑN
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126:135622

DN

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ΤI
     Synthetic catalytic free radical scavengers useful as antioxidants
     for prevention and therapy of disease
     Malfroy-Camine, Bernard; Doctrow, Susan Robin
IN
     Eukarion, Inc., USA; Malfroy-Camine, Bernard; Doctrow, Susan Robin
PA
     PCT Int. Appl., 131 pp.
SO
     CODEN: PIXXD2
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     English
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     MARPAT 126:135622
AΒ
     The invention provides antioxidant salen-metal complexes,
     compns. of such antioxidant salen-metal complexes having
     superoxide activity, catalase activity, and/or peroxidase activity,
     compns. of salen-metal complexes in a form suitable for pharmaceutical
     administration to treat or prevent a disease assocd. with cell or tissue
     damage produced by free radicals such as superoxide, and cosmetic and
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     radical quenching formulations of salen metal compds.
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     RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
         (salen-metal complexes as catalytic free radical scavengers useful as
      antioxidants for prevention and therapy of disease)
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     ICM A61K031-555
     ICS A61K031-28; A61K031-295
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     63-6 (Pharmaceuticals)
     Section cross-reference(s): 1, 62
ST
     salen metal complex radical scavenger antioxidant
IT
     Antioxidants
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Radical scavengers (salen-metal complexes as catalytic free radical scavengers useful as antioxidants for prevention and therapy of disease) 53177-12-1 TΤ RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (salen-metal complexes as catalytic free radical scavengers useful as antioxidants for prevention and therapy of disease) IT 9001-05-2, Catalase 9054-89-1, Superoxide dismutase RL: BSU (Biological study, unclassified); BIOL (Biological study) (salen-metal complexes as catalytic free radical scavengers useful as antioxidants for prevention and therapy of disease) 27815-84-5 51436-86-3 53140-26-4 IT 81065-76-1 156467-55-9 186299-31-0 186299-32-1 186299-33-2 186299-34-3 186299-35-4 186299-36-5 186299-37-6 186299-38-7 186299-39-8 186299-40-1 186299-41-2 186299-42-3 186299-43-4 186299-44-5 186299-45-6 186299-46-7 186358-92-9 RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (salen-metal complexes as catalytic free radical scavengers useful as antioxidants for prevention and therapy of disease) L13 ANSWER 6 OF 7 CAPLUS COPYRIGHT 1999 ACS 1997:28266 CAPLUS AN DN 126:69571 Salen-manganese complexes: combined superoxide dismutase/catalase mimics ΤI with broad pharmacological efficacy Doctrow, Susan R.; Huffman, Karl; Marcus, Catherine B.; Musleh, ΑU Wael; Bruce, Annadora; Baudry, Michel; Malfroy, Bernard CS Eukarion, Inc., Bedford, MA, 01730, USA SO Adv. Pharmacol. (San Diego) (1997), 38 (Antioxidants in Disease Mechanisms and Therapy), 247-269 CODEN: ADPHEL; ISSN: 1054-3589 PB Academic Journal; General Review DT LA English AΒ A review with many refs. The authors describe the catalytic properties of EUK-8, a prototype salen-manganese complex and illustrate its efficacy in exptl. models of disease. Future direction in the development of salen-manganese complexes as novel, broadly applicable potential therapeutic agents. Salen-manganese complexes have several characteristics that might facilitate their potential usefulness as therapeutic agents. First, as low mol. wt., synthetic mols. rather than proteinaceous antioxidant enzymes, they have potential advantages. Second, they activity catalytically, presumably enhancing their over noncatalytic low-mol. wt. ROS scavenger such as vitamin E. Third, their ability to destroy both superoxide anion and hydrogen peroxide should enhance their protective potential in various disease states involving the prodn. of multiple ROS species. **53177-12-1**, EUK-8 RL: BAC (Biological activity or effector, except adverse); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (salen-manganese complexes as combined superoxide dismutase/catalase mimics with broad pharmacol. antioxidant efficacy)

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     Antioxidants (pharmaceutical)
IT
        (salen-manganese complexes as combined superoxide dismutase/catalase
        mimics with broad pharmacol. antioxidant efficacy)
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        (salen-manganese complexes as combined superoxide dismutase/catalase
        mimics with broad pharmacol. antioxidant efficacy)
    ANSWER 7 OF 7 CAPLUS COPYRIGHT 1999 ACS
L13
     1994:570581 CAPLUS
AN
     121:170581
DN
TI
     Synthetic catalytic free radical scavengers useful as antioxidants
     for prevention and therapy of disease
     Malfroy-Camine, bernard; Baudry, Michel
IN
     Eukarion, Inc., USA
PΑ
     PCT Int. Appl., 86 pp.
SO
     CODEN: PIXXD2
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os
GI
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The invention provides antioxidant salen-metal complexes in a
AB
     form suitable for pharmaceutical administration to treat or prevent a
     disease assocd. with cell or tissue damage produced by free radicals such
     as superoxide. A particularly effective compd. is I.
     53177-12-1 149580-33-6 149656-59-7
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        (salen metal complexes, as free radical scavengers, for disease
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     157698-77-6 157698-78-7D, acyl derivs.
     157698-79-8
     RL: BIOL (Biological study)
        (free radical scavenger as antioxidant for disease prevention
        and therapy)
     11062-77-4, Superoxide
     RL: BIOL (Biological study)
        (scavengers for, salen metal complexes as antioxidant, for
        disease prevention and therapy)
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